

Ethyl 4-(4,5-dimethoxy-2-nitrophenyl)-6-methyl-1-phenyl-2-thioxo-1,2,3,4-tetrahydropyrimidine-5-carboxylate

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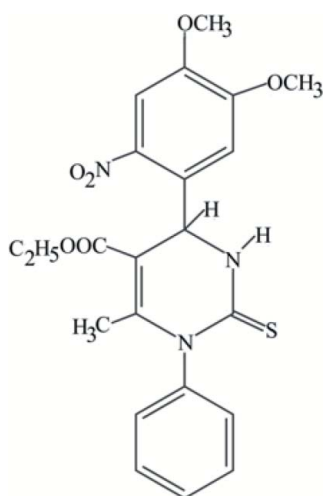
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Key indicators: single-crystal X-ray study; $T = 296$ K; mean $\sigma(\text{C}-\text{C}) = 0.002$ Å; disorder in main residue; R factor = 0.044; wR factor = 0.130; data-to-parameter ratio = 17.3.

In the title compound, $\text{C}_{22}\text{H}_{23}\text{N}_3\text{O}_6\text{S}$, the pyrimidine ring adopts a screw boat conformation, with one C and one N atom displaced from the plane of the remaining atoms. In the crystal structure, centrosymmetric dimers are formed *via* pairs of $\text{N}-\text{H}\cdots\text{S}$ hydrogen bonds. The keto O atom of the ester group is disordered equally over two positions.

Related literature

For related literature, see: Kumaradhas *et al.* (2007).



Experimental

Crystal data

$\text{C}_{22}\text{H}_{23}\text{N}_3\text{O}_6\text{S}$
 $M_r = 457.49$
 Triclinic, $P\bar{1}$
 $a = 8.1475$ (8) Å
 $b = 10.3588$ (10) Å
 $c = 14.8263$ (15) Å
 $\alpha = 86.620$ (2)°
 $\beta = 78.262$ (2)°
 $\gamma = 68.395$ (2)°
 $V = 1138.90$ (19) Å³
 $Z = 2$
 Mo $K\alpha$ radiation
 $\mu = 0.19$ mm⁻¹
 $T = 296$ (2) K
 $0.21 \times 0.11 \times 0.08$ mm

Data collection

Bruker APEX II CCD diffractometer
 Absorption correction: none
 13201 measured reflections
 5229 independent reflections
 4549 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.017$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.044$
 $wR(F^2) = 0.130$
 $S = 1.09$
 5229 reflections
 302 parameters
 H-atom parameters constrained
 $\Delta\rho_{\text{max}} = 0.32$ e Å⁻³
 $\Delta\rho_{\text{min}} = -0.22$ e Å⁻³

Table 1

Hydrogen-bond geometry (Å, °).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
$\text{N2}-\text{H2}\cdots\text{S}^i$	0.86	2.61	3.4505 (11)	167
$\text{C8}-\text{H8B}\cdots\text{O3}^{ii}$	0.96	2.56	3.400 (2)	146

Symmetry codes: (i) $-x + 1, -y + 2, -z + 1$; (ii) $x, y - 1, z$.

Data collection: *APEX2* (Bruker, 2000); cell refinement: *SAINT*; data reduction: *SAINT* (Bruker, 2000); program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *ORTEP-3* (Farrugia, 1997); software used to prepare material for publication: *WinGX* (Farrugia, 1999).

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: HB2430).

References

- Bruker (2000). *APEX2* and *SAINT*. Bruker AXS Inc., Madison, Wisconsin, USA.
 Farrugia, L. J. (1997). *J. Appl. Cryst.* **30**, 565.
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 Kumaradhas, P., David Stephen, A., Satheesh Rai, N., Kalluraya, B. & Sridhar, B. (2007). *Acta Cryst.* **E63**, o3402.
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supplementary materials

Acta Cryst. (2007). E63, o3403 [doi:10.1107/S1600536807031534]

Ethyl 4-(4,5-dimethoxy-2-nitrophenyl)-6-methyl-1-phenyl-2-thioxo-1,2,3,4-tetrahydropyrimidine-5-carboxylate

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Comment

As part of our ongoing studies of biologically active molecules (Kumaradhas *et al.*, 2007), we now report the synthesis and structure of the title compound, (I), (Fig. 1).

Within the 1,4-pyrimidine ring, the C10—N2 and C10—N3 bonds are somewhat short, and the C1—C2 bond of the substituent 2-nitro-4,5-dimethoxy phenyl ring is lengthened, probably due to steric factors. The pyrimidine ring shows a distorted envelope conformation in which C9 and N3 lie above the mean plane of the other ring atoms, having deviations of 0.303 (1) Å and 0.144 (1) Å respectively.

The dihedral angle of the pendant phenyl and aromatic rings with the central ring are 85.9 (1)° and 86.4 (1)° respectively, which marks the near-perpendicularity of both the rings.

In the crystal, the molecules of (I) form inversion dimers (Table 1, Fig. 2) *via* pairs of N—H...S hydrogen bonds thus generating $R_2^2(8)$ loops. A weak C—H...O interaction also occurs.

Experimental

A mixture of 2-nitroveratraldehyde (0.01 mol), ethylaceto acetate (0.015 mole), thiourea (0.01 mol) and concentrated sulfuric acid (1–2 drops) in ethanol (10 ml) taken in a borosil beaker (100 ml capacity) was zapped inside a microwave oven for 3–4 minutes (power = 160 watt). The reaction mixture was then allowed to stand at room temperature and poured into water. The product separated was filtered, washed with water followed by ethanol, dried and finally recrystallized from ethanol by slow evaporation at room temperature to yield pale yellow rectangular blocks of (I).

Refinement

Atom O5 is disordered over two sites with equal occupancies. All the H atoms were placed geometrically (C—H = 0.93–0.97 Å, N—H = 0.86 Å) and refined as riding with $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{carrier})$ or $1.5U_{\text{eq}}(\text{methyl C})$.

Figures

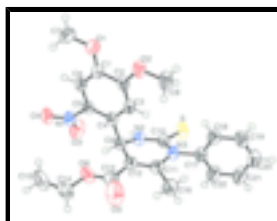


Fig. 1. The molecular structure of (I), showing 50% displacement ellipsoids and arbitrary spheres for the H atoms. Only one orientation of the disordered atom O5 is shown.

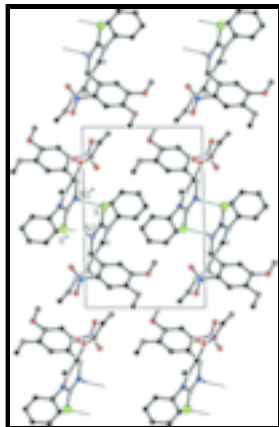


Fig. 2. Part of the crystal structure of (I), showing the formation of an $R_2^2(8)$ dimer. Atoms marked with an asterisk (*) are at the symmetry positions $(1-x, 2-y, 1-z)$. The H atoms not involved in these motifs shown have been omitted.

Ethyl 4-(4,5-dimethoxy-2-nitrophenyl)-6-methyl-1-phenyl-2-thioxo-1,2,3,4-tetrahydropyrimidine-5-carboxylate

Crystal data

$C_{22}H_{23}N_3O_6S$

$M_r = 457.49$

Triclinic, $P\bar{1}$

Hall symbol: $-P\ 1$

$a = 8.1475\ (8)\ \text{\AA}$

$b = 10.3588\ (10)\ \text{\AA}$

$c = 14.8263\ (15)\ \text{\AA}$

$\alpha = 86.620\ (2)^\circ$

$\beta = 78.262\ (2)^\circ$

$\gamma = 68.395\ (2)^\circ$

$V = 1138.90\ (19)\ \text{\AA}^3$

$Z = 2$

$F_{000} = 480$

$D_x = 1.334\ \text{Mg m}^{-3}$

Mo $K\alpha$ radiation

$\lambda = 0.71073\ \text{\AA}$

Cell parameters from 25 reflections

$\theta = 5\text{--}11^\circ$

$\mu = 0.19\ \text{mm}^{-1}$

$T = 296\ (2)\ \text{K}$

Block, pale yellow

$0.21 \times 0.11 \times 0.08\ \text{mm}$

Data collection

Bruker APEX II CCD
diffractometer

ω scans

Absorption correction: none

13201 measured reflections

5229 independent reflections

4549 reflections with $I > 2\sigma(I)$

$R_{\text{int}} = 0.017$

$\theta_{\text{max}} = 28.0^\circ$

$\theta_{\text{min}} = 2.1^\circ$

$h = -10 \rightarrow 10$

$k = -13 \rightarrow 13$

$l = -19 \rightarrow 18$

Refinement

Refinement on F^2

Least-squares matrix: full

$R[F^2 > 2\sigma(F^2)] = 0.044$

H-atom parameters constrained

$$w = 1/[\sigma^2(F_o^2) + (0.0709P)^2 + 0.188P]$$

where $P = (F_o^2 + 2F_c^2)/3$

$(\Delta/\sigma)_{\text{max}} = 0.001$

$wR(F^2) = 0.130$

$S = 1.09$

5229 reflections

302 parameters

$\Delta\rho_{\max} = 0.32 \text{ e } \text{\AA}^{-3}$

$\Delta\rho_{\min} = -0.22 \text{ e } \text{\AA}^{-3}$

Extinction correction: none

Special details

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$	Occ. (<1)
S	0.45829 (6)	0.83000 (4)	0.43847 (3)	0.05704 (14)	
O1	0.66977 (18)	0.55834 (12)	0.92626 (9)	0.0652 (3)	
O2	0.52874 (18)	0.47350 (11)	0.81277 (8)	0.0621 (3)	
O3	0.3863 (2)	1.11302 (13)	0.77587 (9)	0.0747 (4)	
O4	0.38192 (18)	1.07621 (12)	0.92019 (8)	0.0638 (3)	
O5	-0.2304 (13)	0.9803 (10)	0.8041 (8)	0.090 (2)	0.5
O5'	-0.1923 (13)	0.9289 (10)	0.8221 (8)	0.089 (2)	0.5
O6	-0.04308 (15)	1.06751 (13)	0.83484 (8)	0.0618 (3)	
N1	0.39604 (17)	1.03669 (12)	0.84201 (9)	0.0473 (3)	
N2	0.32962 (15)	0.93324 (12)	0.60670 (8)	0.0424 (3)	
H2	0.3986	0.9801	0.5912	0.051*	
N3	0.18386 (16)	0.80205 (12)	0.56207 (8)	0.0447 (3)	
C1	0.56626 (19)	0.65847 (15)	0.87698 (10)	0.0461 (3)	
C2	0.4914 (2)	0.61169 (14)	0.81356 (10)	0.0457 (3)	
C3	0.38765 (19)	0.70637 (14)	0.75925 (9)	0.0434 (3)	
H3	0.3384	0.6746	0.718	0.052*	
C4	0.35371 (17)	0.84891 (13)	0.76403 (8)	0.0372 (3)	
C5	0.42686 (17)	0.89020 (13)	0.82832 (9)	0.0381 (3)	
C6	0.53023 (18)	0.79808 (14)	0.88500 (9)	0.0429 (3)	
H6	0.5748	0.8307	0.9281	0.052*	
C7	0.7410 (3)	0.6026 (2)	0.99266 (16)	0.0827 (6)	
H7A	0.8186	0.6497	0.9626	0.124*	
H7B	0.8086	0.5235	1.0242	0.124*	
H7C	0.6444	0.6649	1.0362	0.124*	
C8	0.4525 (3)	0.41968 (18)	0.75192 (14)	0.0709 (5)	
H8A	0.3236	0.4612	0.7673	0.106*	
H8B	0.4892	0.3208	0.7582	0.106*	
H8C	0.4936	0.441	0.6895	0.106*	
C9	0.23486 (17)	0.94602 (13)	0.70240 (9)	0.0384 (3)	
H9	0.2029	1.0416	0.7236	0.046*	
C10	0.31935 (18)	0.85618 (13)	0.54131 (9)	0.0411 (3)	
C11	0.04369 (18)	0.84849 (14)	0.64043 (10)	0.0437 (3)	
C12	0.06332 (18)	0.92001 (14)	0.70715 (10)	0.0421 (3)	

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C13	-0.0804 (2)	0.97947 (18)	0.78865 (11)	0.0547 (4)
C14	-0.1783 (3)	1.1408 (2)	0.91359 (14)	0.0761 (5)
H14A	-0.2157	1.0754	0.954	0.091*
H14B	-0.2831	1.2066	0.8929	0.091*
C15	-0.0993 (3)	1.2135 (3)	0.96261 (17)	0.0924 (7)
H15A	0.0002	1.1472	0.9857	0.139*
H15B	-0.1884	1.2664	1.0131	0.139*
H15C	-0.0577	1.2748	0.9213	0.139*
C16	-0.1218 (2)	0.8164 (2)	0.63980 (14)	0.0648 (4)
H16A	-0.2048	0.8462	0.6973	0.097*
H16B	-0.0885	0.7182	0.6316	0.097*
H16C	-0.1776	0.8643	0.5902	0.097*
C17	0.1732 (2)	0.71097 (15)	0.49452 (10)	0.0456 (3)
C18	0.2479 (2)	0.56988 (15)	0.50319 (11)	0.0495 (3)
H18	0.3075	0.5323	0.5513	0.059*
C19	0.2334 (2)	0.48365 (17)	0.43937 (13)	0.0602 (4)
H19	0.2839	0.3878	0.4446	0.072*
C20	0.1453 (3)	0.53926 (19)	0.36889 (13)	0.0667 (5)
H20	0.1357	0.4812	0.3264	0.08*
C21	0.0711 (3)	0.6803 (2)	0.36084 (14)	0.0750 (5)
H21	0.0114	0.7178	0.3127	0.09*
C22	0.0843 (3)	0.76721 (18)	0.42366 (13)	0.0653 (5)
H22	0.0336	0.863	0.4182	0.078*

Atomic displacement parameters (\AA^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
S	0.0731 (3)	0.0625 (3)	0.0431 (2)	-0.0406 (2)	0.00476 (17)	-0.00769 (16)
O1	0.0813 (8)	0.0456 (6)	0.0700 (7)	-0.0122 (6)	-0.0376 (6)	-0.0021 (5)
O2	0.0946 (9)	0.0369 (5)	0.0626 (7)	-0.0250 (6)	-0.0302 (6)	-0.0007 (5)
O3	0.1192 (11)	0.0593 (7)	0.0756 (8)	-0.0612 (8)	-0.0346 (8)	0.0162 (6)
O4	0.0815 (8)	0.0499 (6)	0.0617 (7)	-0.0215 (6)	-0.0166 (6)	-0.0194 (5)
O5	0.050 (2)	0.132 (7)	0.093 (5)	-0.047 (4)	0.014 (2)	-0.040 (4)
O5'	0.080 (5)	0.122 (6)	0.079 (4)	-0.071 (5)	0.026 (3)	-0.019 (3)
O6	0.0557 (6)	0.0628 (7)	0.0611 (7)	-0.0246 (5)	0.0126 (5)	-0.0179 (5)
N1	0.0533 (7)	0.0435 (6)	0.0530 (7)	-0.0261 (5)	-0.0100 (5)	-0.0051 (5)
N2	0.0493 (6)	0.0475 (6)	0.0393 (6)	-0.0309 (5)	-0.0028 (5)	0.0008 (4)
N3	0.0509 (6)	0.0445 (6)	0.0470 (6)	-0.0272 (5)	-0.0085 (5)	-0.0019 (5)
C1	0.0484 (7)	0.0447 (7)	0.0446 (7)	-0.0150 (6)	-0.0096 (6)	-0.0055 (6)
C2	0.0574 (8)	0.0368 (7)	0.0449 (7)	-0.0193 (6)	-0.0089 (6)	-0.0048 (5)
C3	0.0526 (7)	0.0403 (7)	0.0425 (7)	-0.0216 (6)	-0.0102 (6)	-0.0056 (5)
C4	0.0393 (6)	0.0387 (6)	0.0356 (6)	-0.0195 (5)	-0.0004 (5)	-0.0044 (5)
C5	0.0388 (6)	0.0379 (6)	0.0398 (6)	-0.0199 (5)	0.0007 (5)	-0.0074 (5)
C6	0.0435 (7)	0.0470 (7)	0.0417 (7)	-0.0202 (6)	-0.0061 (5)	-0.0081 (5)
C7	0.1015 (15)	0.0638 (11)	0.0877 (14)	-0.0146 (10)	-0.0580 (13)	-0.0021 (10)
C8	0.1120 (15)	0.0466 (9)	0.0721 (11)	-0.0417 (10)	-0.0328 (11)	-0.0006 (8)
C9	0.0419 (6)	0.0363 (6)	0.0395 (6)	-0.0198 (5)	-0.0023 (5)	-0.0018 (5)
C10	0.0479 (7)	0.0380 (6)	0.0419 (7)	-0.0216 (6)	-0.0079 (5)	0.0035 (5)

C11	0.0423 (7)	0.0399 (7)	0.0529 (8)	-0.0205 (6)	-0.0086 (6)	0.0044 (5)
C12	0.0402 (7)	0.0411 (7)	0.0471 (7)	-0.0193 (5)	-0.0057 (5)	0.0038 (5)
C13	0.0459 (8)	0.0655 (9)	0.0538 (8)	-0.0257 (7)	-0.0007 (6)	-0.0023 (7)
C14	0.0600 (10)	0.0847 (13)	0.0672 (11)	-0.0180 (10)	0.0148 (9)	-0.0229 (10)
C15	0.0887 (15)	0.0849 (15)	0.0886 (15)	-0.0272 (12)	0.0184 (12)	-0.0394 (12)
C16	0.0544 (9)	0.0724 (11)	0.0808 (12)	-0.0386 (9)	-0.0101 (8)	-0.0065 (9)
C17	0.0536 (8)	0.0428 (7)	0.0486 (7)	-0.0256 (6)	-0.0124 (6)	0.0004 (6)
C18	0.0551 (8)	0.0448 (7)	0.0536 (8)	-0.0226 (6)	-0.0144 (6)	0.0044 (6)
C19	0.0711 (10)	0.0429 (8)	0.0718 (11)	-0.0268 (7)	-0.0126 (8)	-0.0043 (7)
C20	0.0886 (13)	0.0644 (10)	0.0639 (10)	-0.0435 (10)	-0.0193 (9)	-0.0084 (8)
C21	0.1055 (15)	0.0693 (11)	0.0692 (11)	-0.0398 (11)	-0.0457 (11)	0.0091 (9)
C22	0.0893 (13)	0.0454 (8)	0.0726 (11)	-0.0266 (8)	-0.0392 (10)	0.0089 (7)

Geometric parameters (Å, °)

S—C10	1.6733 (14)	C7—H7C	0.96
O1—C1	1.3558 (18)	C8—H8A	0.96
O1—C7	1.414 (2)	C8—H8B	0.96
O2—C2	1.3496 (16)	C8—H8C	0.96
O2—C8	1.433 (2)	C9—C12	1.5053 (17)
O3—N1	1.2195 (17)	C9—H9	0.98
O4—N1	1.2212 (16)	C11—C12	1.3383 (19)
O5—C13	1.194 (10)	C11—C16	1.505 (2)
O5'—C13	1.221 (10)	C12—C13	1.480 (2)
O6—C13	1.325 (2)	C14—C15	1.457 (3)
O6—C14	1.456 (2)	C14—H14A	0.97
N1—C5	1.4635 (16)	C14—H14B	0.97
N2—C10	1.3257 (17)	C15—H15A	0.96
N2—C9	1.4588 (16)	C15—H15B	0.96
N2—H2	0.86	C15—H15C	0.96
N3—C10	1.3874 (17)	C16—H16A	0.96
N3—C11	1.4166 (18)	C16—H16B	0.96
N3—C17	1.4518 (17)	C16—H16C	0.96
C1—C6	1.3730 (19)	C17—C18	1.370 (2)
C1—C2	1.4128 (19)	C17—C22	1.376 (2)
C2—C3	1.378 (2)	C18—C19	1.389 (2)
C3—C4	1.4020 (18)	C18—H18	0.93
C3—H3	0.93	C19—C20	1.367 (3)
C4—C5	1.3833 (17)	C19—H19	0.93
C4—C9	1.5228 (18)	C20—C21	1.369 (3)
C5—C6	1.388 (2)	C20—H20	0.93
C6—H6	0.93	C21—C22	1.379 (2)
C7—H7A	0.96	C21—H21	0.93
C7—H7B	0.96	C22—H22	0.93
C1—O1—C7	116.94 (13)	N3—C10—S	122.04 (10)
C2—O2—C8	117.96 (12)	C12—C11—N3	119.26 (12)
C13—O6—C14	117.29 (14)	C12—C11—C16	124.83 (14)
O3—N1—O4	123.17 (13)	N3—C11—C16	115.88 (13)
O3—N1—C5	119.24 (12)	C11—C12—C13	122.83 (13)

supplementary materials

O4—N1—C5	117.60 (12)	C11—C12—C9	121.15 (12)
C10—N2—C9	126.12 (11)	C13—C12—C9	116.02 (12)
C10—N2—H2	116.9	O5—C13—O5'	28.2 (6)
C9—N2—H2	116.9	O5—C13—O6	118.7 (6)
C10—N3—C11	121.49 (11)	O5'—C13—O6	122.6 (6)
C10—N3—C17	118.25 (12)	O5—C13—C12	127.9 (6)
C11—N3—C17	119.66 (11)	O5'—C13—C12	123.7 (6)
O1—C1—C6	124.99 (13)	O6—C13—C12	111.26 (12)
O1—C1—C2	115.74 (12)	O6—C14—C15	108.14 (16)
C6—C1—C2	119.27 (13)	O6—C14—H14A	110.1
O2—C2—C3	125.40 (12)	C15—C14—H14A	110.1
O2—C2—C1	115.11 (13)	O6—C14—H14B	110.1
C3—C2—C1	119.49 (12)	C15—C14—H14B	110.1
C2—C3—C4	122.31 (12)	H14A—C14—H14B	108.4
C2—C3—H3	118.8	C14—C15—H15A	109.5
C4—C3—H3	118.8	C14—C15—H15B	109.5
C5—C4—C3	116.15 (12)	H15A—C15—H15B	109.5
C5—C4—C9	125.22 (11)	C14—C15—H15C	109.5
C3—C4—C9	118.58 (11)	H15A—C15—H15C	109.5
C4—C5—C6	123.17 (12)	H15B—C15—H15C	109.5
C4—C5—N1	121.33 (12)	C11—C16—H16A	109.5
C6—C5—N1	115.50 (11)	C11—C16—H16B	109.5
C1—C6—C5	119.56 (12)	H16A—C16—H16B	109.5
C1—C6—H6	120.2	C11—C16—H16C	109.5
C5—C6—H6	120.2	H16A—C16—H16C	109.5
O1—C7—H7A	109.5	H16B—C16—H16C	109.5
O1—C7—H7B	109.5	C18—C17—C22	120.62 (14)
H7A—C7—H7B	109.5	C18—C17—N3	119.71 (13)
O1—C7—H7C	109.5	C22—C17—N3	119.65 (13)
H7A—C7—H7C	109.5	C17—C18—C19	119.26 (14)
H7B—C7—H7C	109.5	C17—C18—H18	120.4
O2—C8—H8A	109.5	C19—C18—H18	120.4
O2—C8—H8B	109.5	C20—C19—C18	120.23 (15)
H8A—C8—H8B	109.5	C20—C19—H19	119.9
O2—C8—H8C	109.5	C18—C19—H19	119.9
H8A—C8—H8C	109.5	C19—C20—C21	120.08 (16)
H8B—C8—H8C	109.5	C19—C20—H20	120
N2—C9—C12	109.01 (10)	C21—C20—H20	120
N2—C9—C4	111.33 (11)	C20—C21—C22	120.35 (16)
C12—C9—C4	112.06 (10)	C20—C21—H21	119.8
N2—C9—H9	108.1	C22—C21—H21	119.8
C12—C9—H9	108.1	C17—C22—C21	119.46 (16)
C4—C9—H9	108.1	C17—C22—H22	120.3
N2—C10—N3	116.24 (12)	C21—C22—H22	120.3
N2—C10—S	121.69 (10)		
C7—O1—C1—C6	1.9 (2)	C17—N3—C10—S	5.02 (18)
C7—O1—C1—C2	-177.74 (17)	C10—N3—C11—C12	-16.1 (2)
C8—O2—C2—C3	-1.2 (2)	C17—N3—C11—C12	172.90 (12)
C8—O2—C2—C1	178.20 (15)	C10—N3—C11—C16	161.79 (14)

O1—C1—C2—O2	2.0 (2)	C17—N3—C11—C16	-9.19 (19)
C6—C1—C2—O2	-177.72 (13)	N3—C11—C12—C13	176.31 (13)
O1—C1—C2—C3	-178.59 (14)	C16—C11—C12—C13	-1.4 (2)
C6—C1—C2—C3	1.7 (2)	N3—C11—C12—C9	-2.9 (2)
O2—C2—C3—C4	179.72 (14)	C16—C11—C12—C9	179.39 (14)
C1—C2—C3—C4	0.3 (2)	N2—C9—C12—C11	21.97 (17)
C2—C3—C4—C5	-1.5 (2)	C4—C9—C12—C11	-101.72 (14)
C2—C3—C4—C9	-178.99 (13)	N2—C9—C12—C13	-157.29 (12)
C3—C4—C5—C6	0.75 (19)	C4—C9—C12—C13	79.02 (15)
C9—C4—C5—C6	178.01 (12)	C14—O6—C13—O5	11.5 (5)
C3—C4—C5—N1	-177.98 (12)	C14—O6—C13—O5'	-21.1 (5)
C9—C4—C5—N1	-0.71 (19)	C14—O6—C13—C12	176.21 (15)
O3—N1—C5—C4	-35.34 (19)	C11—C12—C13—O5	-4.6 (5)
O4—N1—C5—C4	144.93 (13)	C9—C12—C13—O5	174.6 (5)
O3—N1—C5—C6	145.85 (14)	C11—C12—C13—O5'	29.9 (6)
O4—N1—C5—C6	-33.89 (17)	C9—C12—C13—O5'	-150.8 (5)
O1—C1—C6—C5	177.86 (14)	C11—C12—C13—O6	-167.62 (14)
C2—C1—C6—C5	-2.5 (2)	C9—C12—C13—O6	11.62 (19)
C4—C5—C6—C1	1.3 (2)	C13—O6—C14—C15	170.88 (19)
N1—C5—C6—C1	-179.94 (12)	C10—N3—C17—C18	97.45 (16)
C10—N2—C9—C12	-27.44 (18)	C11—N3—C17—C18	-91.28 (17)
C10—N2—C9—C4	96.69 (15)	C10—N3—C17—C22	-84.14 (19)
C5—C4—C9—N2	108.95 (13)	C11—N3—C17—C22	87.13 (18)
C3—C4—C9—N2	-73.85 (14)	C22—C17—C18—C19	0.2 (2)
C5—C4—C9—C12	-128.66 (13)	N3—C17—C18—C19	178.56 (14)
C3—C4—C9—C12	48.54 (16)	C17—C18—C19—C20	-0.1 (3)
C9—N2—C10—N3	11.6 (2)	C18—C19—C20—C21	0.1 (3)
C9—N2—C10—S	-170.21 (11)	C19—C20—C21—C22	-0.1 (3)
C11—N3—C10—N2	12.12 (19)	C18—C17—C22—C21	-0.2 (3)
C17—N3—C10—N2	-176.78 (12)	N3—C17—C22—C21	-178.56 (17)
C11—N3—C10—S	-166.08 (10)	C20—C21—C22—C17	0.1 (3)

Hydrogen-bond geometry (Å, °)

<i>D</i> —H... <i>A</i>	<i>D</i> —H	H... <i>A</i>	<i>D</i> ... <i>A</i>	<i>D</i> —H... <i>A</i>
N2—H2...S ⁱ	0.86	2.61	3.4505 (11)	167
C8—H8B...O3 ⁱⁱ	0.96	2.56	3.400 (2)	146

Symmetry codes: (i) $-x+1, -y+2, -z+1$; (ii) $x, y-1, z$.

Fig. 1

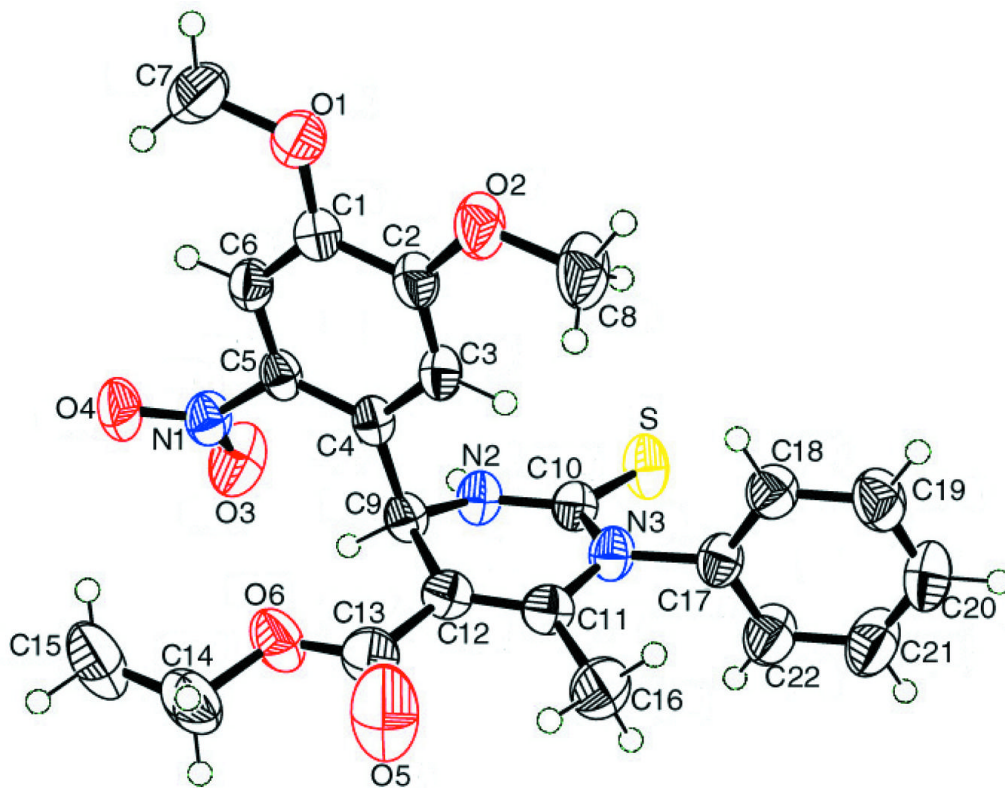


Fig. 2

